

40 C.F.R. § 1065.1005

Symbols, abbreviations, acronyms, and units of measure.

The procedures in this part generally follow the International System of Units (SI), as detailed in NIST Special Publication 811, which we incorporate by reference in § 1065.1010. See § 1065.20 for specific provisions related to these conventions. This section summarizes the way we use symbols, units of measure, and other abbreviations.

(a) *Symbols for quantities.* This part uses the following symbols and units of measure for various quantities:

Table 1 of § 1065.1005—Symbols for Quantities

Symbol	Quantity	Unit	Unit symbol	Units in terms of SI base units
α	atomic hydrogen-to-carbon ratio	mole per mole	mol/mol	1.
A	area	square meter	m ²	m ² .
a ₀	intercept of least squares regression			
a ₁	slope of least squares regression			
ag	acceleration of Earth's gravity	meter per square second	m/s ²	m · s ⁻² .
β	ratio of diameters	meter per meter	m/m	1.
β	atomic oxygen-to-carbon ratio	mole per mole	mol/mol	1.
C#	number of carbon atoms in a molecule			
c	power-specific carbon mass error coefficient	gram per kilowatt-hour	g/(kW·hr)	$3.6 \cdot 10^{-9} \cdot \text{m}^{-2} \cdot \text{s}^2$.
Cd	discharge coefficient			
Cf	flow coefficient			
δ	atomic nitrogen-to-carbon ratio	mole per mole	mol/mol	1.
d	diameter	meter	m	m.
d	power-specific carbon mass rate absolute error coefficient	gram per kilowatt-hour	g/(kW·hr)	$3.6 \cdot 10^{-9} \cdot \text{m}^{-2} \cdot \text{s}^2$.
DR	dilution ratio	mole per mole	mol/mol	1.

ϵ	error between a quantity and its reference			
\in	difference or error quantity			
e	brake-specific emission or fuel consumption	gram per kilowatt hour	g/(kW·hr)	$3,6 \cdot 10^{-9} \cdot m^{-2} \cdot s^2$.
F	F-test statistic			
f	frequency	hertz	Hz	s^{-1} .
fn	angular speed (shaft)	revolutions per minute	r/min	$\pi \cdot 30^{-1} \cdot s^{-1}$.
γ	ratio of specific heats	(joule per kilogram kelvin) per (joule per kilogram kelvin)	(J/(kg·K))/(J/(kg·K))	1.
γ	atomic sulfur-to-carbon ratio	mole per mole	mol/mol	1.
κ	opacity			
K	correction factor			1.
Kv	calibration coefficient		$m^4 \cdot s \cdot K^{0.5}/kg$	$m^4 \cdot kg^{-1} \cdot s \cdot K^{0.5}$.
l	length	meter	m	m.
L	limit			
μ	viscosity, dynamic	pascal second	Pa·s	$m^{-1} \cdot kg \cdot s^{-1}$.
M	molar mass 1	gram per mole	g/mol	$10^{-3} \cdot kg \cdot mol^{-1}$.
m	mass	kilogram	kg	kg.
m	mass rate	kilogram per second	kg/s	$kg \cdot s^{-1}$.
ν	viscosity, kinematic	meter squared per second	m ² /s	$m^2 \cdot s^{-1}$.
N	total number in series			
n	amount of substance	mole	mol	mol.
n	amount of substance rate	mole per second	mol/s	$mol \cdot s^{-1}$.
P	power	kilowatt	kW	$10^3 \cdot m^2 \cdot kg \cdot s^{-3}$.
PF	penetration fraction			
p	pressure	pascal	Pa	$m^{-1} \cdot kg \cdot s^{-2}$.
ρ	mass density	kilogram per cubic meter	kg/m ³	$m^{-3} \cdot kg$.
Δp	differential static pressure	pascal	Pa	$m^{-1} \cdot kg \cdot s^{-2}$.
r	ratio of pressures	pascal per pascal	Pa/Pa	1.

r^2	coefficient of determination			
R_a	average surface roughness	micrometer	μm	$10^{-6} \cdot \text{m}$.
$Re\#$	Reynolds number			
RF	response factor			
RH	relative humidity			
σ	non-biased standard deviation			
S	Sutherland constant	kelvin	K	K.
SEE	standard error of the estimate			
T	absolute temperature	kelvin	K	K.
T	Celsius temperature	degree Celsius	$^{\circ}\text{C}$	$\text{K} - 273.15$.
T	torque (moment of force)	newton meter	N·m	$\text{m}^2 \cdot \text{kg} \cdot \text{s}^{-2}$.
θ	plane angle	degrees	$^{\circ}$	rad.
t	time	second	s	s.
Δt	time interval, period, 1/frequency	second	s	s.
V	volume	cubic meter	m^3	m^3 .
V	volume rate	cubic meter per second	m^3/s	$\text{m}^3 \cdot \text{s}^{-1}$.
W	work	kilowatt-hour	kWhr	$3.6 \cdot 10^6 \cdot \text{m}^2 \cdot \text{kg} \cdot \text{s}^{-2}$.
w_C	carbon mass fraction	gram per gram	g/g	1.
x	amount of substance mole fraction. ²	mole per mole	mol/mol	1.
$X \sim$	flow-weighted mean concentration	mole per mole	mol/mol	1.
y	generic variable			
Z	compressibility factor			

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