
40 C.F.R. § 60.1460

What equations must I use?

(a) *Concentration correction to 7 percent oxygen.* Correct any pollutant concentration to 7 percent oxygen using equation 1 of this section:

$$C_{7\%} = C_{\text{unc}} * (13.9) * (1/(20.9 - \text{CO}_2)) \text{ (Eq.1)}$$

Where:

$C_{7\%}$ = concentration corrected to 7 percent oxygen. C_{unc} = uncorrected pollutant concentration. CO_2 = concentration of oxygen (percent).

(b) *Percent reduction in potential mercury emissions.* Calculate the percent reduction in potential mercury emissions ($\%P_{\text{Hg}}$) using equation 2 of this section:

$$\%P_{\text{Hg}} = (E_{i-o}) * (100/E_i) \text{ (Eq. 2)}$$

Where:

$\%P_{\text{Hg}}$ = percent reduction of potential mercury emissions E_i = mercury emission concentration as measured at the air pollution control device inlet, corrected to 7 percent oxygen, dry basis E_o = mercury emission concentration as measured at the air pollution control device outlet, corrected to 7 percent oxygen, dry basis

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